AMENDMENTS TO THE SPECIFICATION:

Please insert the following headings on page 1, line 3:

BACKGROUND OF THE INVENTION

Field of the invention:

Please insert the following heading on page 1, line 9:

Description of the Related Art

Please insert the following heading on page 3, line 5:

SUMMARY OF THE INVENTION

Please insert the following heading on page 3, line 20:

DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENTS

Please amend the paragraph on page 4, starting at line 20 and ending on line 33:

The phosphonate compounds corresponding to the formula (II) in which z = 4 and R is an alkyl Ra can be prepared, for example, by the following process:

Attorney's Docket No. <u>032013-116</u> Application No. Unassigned Page 3

- during a first stage, the trialkyl phosphite P(ORa)₃ (VI) is reacted with the dibromoalkylene dibromoalkane Br-(CH₂)_y-Br (VII) at a temperature of the order of 140°C in order to obtain Br-(CH₂)_y-P(O)(ORa)₂ (VIII),
- during a second stage, the phosphonate Br-(CH₂)_y-P(O) (ORa)₂ (VIII) is reacted with Na₂S₄ under reflux of the methanol. A product is obtained having a mean composition corresponding to the formula

$$(RaO)_2P(O)-(CH_2)_v-S_4-(CH_2)_v-P(O) (ORa)_2$$
 (IIa)

in which Ra is an alkyl.

Please amend the paragraph beginning on page 5 at line 17 and ending on line 29:

A phosphate (IIIa) corresponding to the mean formula (III) in which R is H can be obtained by a process in which:

- during a first stage, P(O)Cl₃ is reacted with a compound HO(CH₂)_yCl in stoichiometric proportions in order to obtain the compound Cl(CH₂)_yOP(O)Cl₂;
- during a second stage, the compound CL(CH₂)_yOP(O)Cl₂ is hydrolyzed in order to obtain the compound Cl(CH₂)_yOPO₃H₂;
- during a third stage, Cl(CH₂)_yOPO₃H₂ is reacted with Na₂S₄ under reflux of the methanol and then an ion exchange is carried out in order to obtain the compound

Attorney's Docket No. <u>032013-116</u> Application No. Unassigned Page 4

$$\begin{split} & [[(HO)_2P(O)-O-(CH_2)_y-S_z-(CH_2)_y-O-P(O)(OH)_2]] \ \underline{(HO)_2P(O)-O-(CH_2)_yS_4-(CH_2)_y-O-P(O)(OH)_2} \\ & \underline{P(O)(OH)_2}. \end{split}$$